

Long Term Plan – Triple Science (2024-25)

Vision stude know scient globa	a: Our vision is to ignite a pas nts are empowered to explo ledge to address real-world o ific landscape. We are comm l community.	ssion for science in every s re the wonders of science challenges. By integrating nitted to fostering a cultur	student. We strive to created on the strive to created on the strive to created on the sustainable practices and the scientific curiosity and the scientific curi	ate a dynamic, hands-on l ritical thinkers and proble d collaborative projects, w nd inspiring students to m	earning environment whe m solvers equipped with ve prepare students to lea nake meaningful contribu	ere curiosity thrives and the skills and ad in a rapidly evolving tions to society and the	
	HT1 Sept – Oct	HT2 - Oct-Dec	HT3 - Jan-Feb	HT4 - Feb-Mar	HT5 - Apr-May	HT6 - June-July	-
Year 10 Biology	B4 Organising animals and plants, blood, gas exchange and transport systems in plants.	B5 Communicable disease triple only content. B6, B7 non- communicable disease and the prevention of disease. Growing bacteria required practical.	B8 photosynthesis Required practical Photosynthesis.	B9 respiration B10 Nervous system Required practical Reaction time.	B10 Nervous system	B16 Adaptation, interdependence and competition Required practical – Field investigations	By the end of year 10 students The need for transport system relationship between the struct and the function of the gas exec The relationship between heal sexually transmitted infections diseases , the role of bacteria, the body's defences against padise disease, how to reduce and pro- plants, the process of the disco impact of lifestyle factors on the How monoclonal antibodies ar pregnancy tests, the diagnosis disease and research. The process of photosynthesis The importance of cellular resp respiration. The principles of nervous coordination between the structure and function Some abiotic and biotic factors interactions between organism measuring distribution, freque organisms are interdependent
Year 10 Chemistry	C2.6 Transition metals, C3.11Nanoparticles, C3.12 Applications of nanoparticle need to be covered as triple only content. C5 chemical changes	C5 chemical changes Required practical Making salts. C6 Electrolysis Required practical Electrolysis.	C4 – Chemical calculations Required practical Neutralisation.	C4 – Chemical calculations Required practical Neutralisation.	C7 Energy changes Required practical Temperature changes.	C8 Rates and equilibrium and required practical (rates.) Required practical Rates of reaction.	The different chemical change be classified in different ways. substances will be formed. That the extraction of importa some elements and compound 'pulled apart' using the princip How to represent chemical real use quantitative methods to d

Year End Points

will know:

ns in multicellular organisms, including plants, the cture and functions of the human circulatory system in animals.

Ith and disease, communicable diseases including is in humans (including HIV/AIDs), non-communicable viruses and fungi as pathogens in animals and plants, athogens and the role of the immune system against revent the spread of infectious diseases in animals and overy and development of new medicines and the the incidence of non-communicable diseases.

re made and some examples of their uses including s of disease, monitoring levels of infection, treating

and the factors affecting the rate of photosynthesis.

piration and the processes of aerobic and anaerobic

rdination and control in humans, the relationship nction of the human nervous system, the relationship on in a reflex arc.

rs which affect communities; the importance of ms in a community, methods of identifying species and ency and abundance of species within a habitat, t and are adapted to their environment. s will know:

es that can take place and these chemical changes can . Students will be able to predict exactly which new

ant resources from the earth makes use of the way that ds react with each other and how easily they can be ples of electrolysis.

actions and given information, students will be able to determine the purity of chemical samples.



							How to use titration to determ That energy changes are an im of particles often involves the t bonds. Reactions in which ener reactions, while those that takk know that these interactions b that are used in a range of eve ions in an electrolyte result in t Cells and batteries use these ch also know that electricity can a useful means of producing eler That chemical reactions can oc chemicals is a significant factor many variables that can be ma down. Chemical reactions may variables needs to be establish desired product. Students will determine the effect of differe whilst there may be compromi to ensure that enough product efficient way.
Year 10 Physics	P3 Energy resources P5 Electricity in the home	P6 Molecules and matter including P2.2, P2.3, Required practical Thermal insulation Required practicals Specific heat capacity and density	P7 Radioactivity P9 Motion	P8 Complete Forces in balance	P10 Forces and motion Required practicals Acceleration and Force and extension.	P11 Force and Pressure Paper 1 retrieval in preparation for the mock exam	By the end of year 10 students Renewable and non-renewable are used. Power as the rate of transfer of transfers, the domestic a.c. sup measures, power transfer rela Relating models of arrangemen phases to their densities, melti calculating energy changes inve- those involved in changes of st and temperature of a gas at co- (qualitative). The nuclear model and its deve- sizes of nuclei, atoms and small neutrons related to masses and equations to represent change related to changes in electron particles, neutrons, or gamma charge, radioactive materials, h hazardous effects, waste dispo- energy. Speed of sound, estimating spe- interpreting quantitatively grap

ine the concentration of acids and alkalis.

portant part of chemical reactions and the interaction transfer of energy due to the breaking and formation of rgy is released to the surroundings are exothermic te in thermal energy are endothermic. Students will between particles can produce heating or cooling effects eryday applications and that some interactions between the production of electricity.

hemical reactions to provide electricity. Students will also be used to decompose ionic substances and is a ments that are too expensive to extract any other way.

ccur at vastly different rates. Whilst the reactivity of r in how fast chemical reactions proceed, there are anipulated in order to speed them up or slow them y also be reversible and therefore the effect of different ned in order to identify how to maximise the yield of know that in industry, chemists and chemical engineers ent variables on reaction rate and yield of product and ises to be made, they carry out optimisation processes t is produced within a sufficient time, and in an energy-

will know:

e energy sources used on Earth, changes in how these

of energy, calculating energy efficiency for any energy pply; live, neutral and earth mains wires, safety ated to p.d. and current, or current and resistance.

nts and motions of the molecules in solid, liquid and gas ing, evaporation, and sublimation as reversible changes, volved on heating, using specific heat capacity; and tate, using specific latent heat, links between pressure onstant volume, related to the motion of its particles

elopment in the light of changing evidence, masses and ill molecules, differences in numbers of protons, and id identities of nuclei, isotope characteristics and es, ionisation; absorption or emission of radiation orbits, radioactive nuclei: emission of alpha or beta rays, related to changes in the nuclear mass and/or half-life, irradiation, contamination and their associated osal, nuclear fission, nuclear fusion and our Sun's

eeds and accelerations in everyday contexts, phs of distance, time, and speed.



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						Forces and fields: electrostatic done as force x distance; elast directions: variation in Earth's thrust force (qualitative). Acceleration caused by forces; strength, decelerations and br Pressure, changes in pressure atmosphere, upthrust, floating
Year 11 Biology	B11 Hormonal control Required practical Plant responses.	B12 Homeostasis in action B13 Reproduction.	B14 Variation and evolution	B15 Genetics and evolution	B18 Biodiversity and ecosystems.	By the end of year 11 students The principles of homeostasis maintains optimum conditions of the human nervous system, reflex arc. The principles of hormonal con- human reproduction and the of contraception. How plant horn how hormones are used in ho The detail of homeostasis inclu- using the hormone ADH. The b the blood and controlling wate The genome as the entire gen- its interaction with the environ sex is determined in humans. How single gene inheritance a phenotypes can occur. The ide many genes and how genomic screening and the ethics involv That there is genetic variety w leads to evolution. The eviden classification has developed. T Baptiste Lamarck, Charles Dar The importance of selective be the use of modern biotechnolo- challenges. How materials cycle through a role of microorganisms (decor- ecosystem. How organisms ar and the importance of biodive interactions with ecosystems.

c, magnetic, gravity, forces as vectors, calculating work tic and inelastic stretching, pressure in fluids acts in all atmosphere with height, with depth for liquids, up-

; Newton's First Law, weight and gravitational field raking distances involved on roads, safety.

with variation in depth in water and also in the g and density.

s will know:

and how nervous coordination and control in humans s. The relationship between the structure and function , the relationship between structure and function in a

oordination in humans including the hormones involve in use of hormonal and non-hormonal methods of mones enable plants to respond to light and gravity and orticulture.

uding the control of body temperature and water levels basic structure of the kidney and its function in filtering er levels in the body.

etic material of an organism and how the genome and nment influence the phenotype of an organism. How

and single gene crosses involving dominant and recessive ea that most phenotypes result from the interaction of cs can impact medicine through ideas such as embryo lved.

vithin populations of a species and natural selection nee that scientists use for evolution and how the study of The role of key scientists including; Gregor Mendel, Jean twin, Alfred Wallace Carl Linnaeus. including;

reeding of both plants and animals in agriculture and ogy in farming along with the practical and ethical

abiotic and biotic components of ecosystems and the mposers) in the cycling of materials through an re interdependent and are adapted to their environment ersity and some of the positive and negative human



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Year 11 Chemistry	C12 Chemical analysis Required practical Chromatography and Identifying ions. C9 Crude oil	C10 Organic reactions C11 Polymers	C13 The Earth's atmosphere C14 The Earth's resources Required practical Water purification.	C15 Using our resources.		By the end of year 11 student How to distinguish between p separation techniques for mix chromatography, simple and t That carbon compounds are u demands for limited resource are used to make more useful How important organic chemi are possible due carbon comp How to identify different type how to name these compound polymerisation in creating use How to apply life cycle assess associated with all the stages The evidence for the composi its formation. The evidence, a The potential effects of, and r methane on the Earth's clima dioxide, oxides of nitrogen, pa The Earth's water resources th we obtain potable water and into the water courses.
Year 11 Physics	P12 Wave properties Required practical Waves.	P13 Electromagnetic waves Required practical Radiation and absorption. P14 Light Required practical light.	P15 Electromagnetism	P16 Space		By the end of year 11 student Amplitude, wavelength, freque transverse and longitudinal we reflection, refraction effects Electromagnetic waves, veloce and frequencies from radio to circuits, or by changes in atom visible, ultra-violet, X-ray and Reflection and refraction, the depends on the medium (tran- it passes through different len Exploring the magnetic fields magnetic field, using a compa- the effect, how transformers a use. The life cycle of a star, how re- method for the creation of the

ts will know:

pure and impure substances and how to carry out xtures of substances: filtration, crystallisation, fractional distillation.

used both as fuels and feedstock, and the competing es. How fractional distillation of crude oil and cracking Il materials.

histry is and the great variety of carbon compounds that pounds forming chains and rings linked by carbon atoms. es of organic molecules by their functional group and hds. The use of addition and condensation heful products.

sment and recycling to assess environmental impacts s of a product's life.

sition and the evolution of the Earth's atmosphere since and the uncertainties in evidence, for climate change. mitigation of, increased levels of carbon dioxide and ate. Other common atmospheric pollutants: sulphur particulates and their sources.

through the detailed study of the water cycle and how the treatment of waste water to allow safe discharge

ts will know:

uency, relating velocity to frequency and wavelength, vaves, velocities differing between media: absorption,

city in vacuum; waves transferring energy; wavelengths o gamma-rays, production and detection, by electrical ms and nuclei, uses in the radio, microwave, infra-red, I gamma ray regions, hazardous effects on bodily tissues.

e causes of refraction, coloured light, how light behaves nsparent, translucent, opaque), what happens to light as nses.

of permanent and induced magnets, and the Earth's ass, magnetic effects of currents, how solenoids enhance are used in the national grid and the reasons for their

ed-shift has proven the Big Bang to be the most realistic the Universe, the main features of the solar system.